

Pesticide Transport and Reaction in Soil

Introduction

Aldicarb is a commercial pesticide, used on a variety of crops, including cotton, sugar beet, citrus fruits, potatoes, and beans. The general population may be exposed to aldicarb primarily through the ingestion of contaminated water and foods.

This example looks at the degradation kinetics of aldicarb and its toxic by-products, investigating both the degradation time-scale as well as the spatial concentration distribution of toxic components. In the first model the chemicals are contained in a water pond, treated as a perfectly mixed system. The second model tracks the detailed distribution of chemicals in soil as the pesticide leaches out of the pond and is transported in water through the ground.

Model Definition

Aldicarb degrades by transformation to the corresponding sulfoxide and the sulfone (both of which are toxic), and is detoxified by hydrolysis to oximes and nitriles. The chain of reactions is illustrated in Figure 1. The toxicity of a chemical species is indicated by its LD_{50} value, signifying the median lethal dose (mg/kg) to half of a test population of rats. As indicated, both the sulfoxide and sulfone analogues of aldicarb are also relatively toxic.



Figure 1: Reaction pathways of aldicarb degradation.

Each of the j unimolecular reactions outlined above has a rate expression of the form

$$r_j = k_j c_i$$

Note that in this example the concentration unit is mol/m^3 and the rate constants are expressed in 1/day.

PERFECTLY MIXED SYSTEM

The first model solves for the decomposition kinetics of aldicarb occurring in a water pond. The pond is treated as a closed and perfectly mixed system. The reaction mechanism illustrated in Figure 1 translates into the following mass balance equations:

• For aldicarb (a)

$$\frac{dc_a}{dt} = -r_1 - r_3$$

• For aldicarb sulfoxide (asx)

$$\frac{dc_{\text{asx}}}{dt} = r_1 - r_2 - r_4$$

• For aldicarb sulfone (asn)

$$\frac{dc_{\rm asn}}{dt} = r_2 - r_5$$

• For aldicarb oxime (ao)

$$\frac{dc_{ao}}{dt} = r_3$$

• For aldicarb sulfoxide oxime (asxo)

$$\frac{dc_{asxo}}{dt} = r_4$$

• For aldicarb sulfone oxime (asno)

$$\frac{dc_{\rm asno}}{dt} = r_5$$

Solving this set of coupled ODEs provides information on the time scales of the degradation processes.

SPACE- AND TIME-DEPENDENT SYSTEM

In a more detailed model, assume that aldicarb moves from the pond into relatively dry soil. In the soil, the aldicarb decomposes according to the mechanism illustrated in Figure 1. In addition, the pesticide and its decay products are transported by convection, dispersion, sorption, and volatilization.

Geometry

Water is ponded by a ring sitting on the ground. The soil is layered and rests on rocks, with the top layer slightly less permeable than the bottom one. Water moves through the bottom of the ring into the soil. The water level in the ring is known, as is the initial distribution of pressure heads in the soil. There is no flow through the vertical walls or the surface outside of the ring.



Figure 2: Geometry of the infiltration ring and soil column.

Aldicarb moves with water from the pond into the soil at a constant concentration. In the soil, the chemicals react and also adsorb onto soil particles. Aldicarb and the aldicarb sulfone volatilize to the atmosphere. The sorption, biodegradation, and volatilization proceed in linear proportion to the aqueous concentrations. The soil is initially pristine with zero concentration of the involved chemicals. At the ground surface outside the ring there is volatilization to the atmosphere for c_a and c_{asn} . Model the problem with 2D axisymmetry and track the solute transport for 10 days. The vertical axis to the left is the symmetry axis. Add infinite elements to the right axis such that the solutes can freely leave the soil column with the fluid flow.

Fluid Flow

Richards' equation governs the saturated-unsaturated flow of water in the soil. The soil pores are connected to the atmosphere, so you can assume that pressure changes in the air do not affect the flow and use Richards' equation for single-phase flow. Given by Ref. 1, Richards' equation in pressure head reads

$$(C_{\rm m} + S_{\rm e}S_{\rm p})\frac{\partial H_p}{\partial t} + \nabla \cdot (-K\nabla(H_p + D)) = 0$$

where $C_{\rm m}$ denotes specific moisture capacity (m⁻¹); S_e is the effective saturation of the soil (dimensionless); S_p is a storage coefficient (m⁻¹); H_p is the pressure head (m), which is proportional to the dependent variable, p (Pa); t is time; K equals the hydraulic conductivity (m/s); D is the direction (typically, the z direction) that represents vertical elevation (m).

To be able to combine boundary conditions and sources with the Darcy's Law formulation, COMSOL Multiphysics converts Richards' equation to SI units and solves for the pressure (SI unit: Pa). Hydraulic head, H, pressure head, H_p , and elevation D are related to pressure p as

$$H_p = \frac{p}{\rho g}; \qquad H = H_p + D$$

Also, the permeability κ (SI unit: $1/m^2$) and hydraulic conductivity K (SI unit: m/s) are related to the viscosity μ (SI unit: Pa·s) and density ρ (SI unit: kg/m³) of the fluid, and the acceleration of gravity g (SI unit: m/s^2) by

$$\frac{\kappa}{\mu} = \frac{K}{\rho g}$$

In this problem, $S_p = (\theta_s - \theta_r)/(1 \text{ m} \cdot \rho g)$ where θ_s and θ_r denote the volume fraction of fluid at saturation and after drainage, respectively. For more details see *The Richards' Equation Interface* in the *Subsurface Flow Module User's Guide*.

Mass Transport

The governing equation for solute transport describes advection and dispersion of a sorbing, volatilizing, and decaying solute in variably saturated soil.

$$\frac{\partial}{\partial t}(\theta c) + \frac{\partial}{\partial t}(\rho_{\rm b}c_{\rm P}) + \mathbf{u} \cdot \nabla c + \nabla \cdot (-\theta D_{\rm L} \nabla c) = \Sigma R_{\rm L} + \Sigma R_{\rm P} + S_{\rm c}$$
(1)

The Transport of Diluted Species in Porous Media interface implements Equation 1. It describes the time rate of change in two terms: c denotes dissolved concentration (mol/

 m^3) and c_P is the mass of adsorbed contaminant per dry unit weight of solid (mg/kg). Further, θ denotes the volume fraction of fluid (porosity), and ρ_b is the bulk density (kg/ m^3). Because ρ_b amounts to the dried solid mass per bulk volume of the solids and pores together, the term $\rho_b c_P$ gives solute mass attached to the soil as the concentration changes with time.

Solute spreading now includes mechanical dispersion in water plus molecular diffusion for water and air. These three processes appear in the liquid-gas dispersion tensor, whose entries are

$$\begin{split} \theta D_{\mathrm{LG}ii} &= \alpha_1 \frac{u_i^2}{|\mathbf{u}|} + \alpha_2 \frac{u_j^2}{|\mathbf{u}|} + \theta \frac{D_{\mathrm{m}}}{\tau_{\mathrm{L}}} + \alpha_{\mathrm{v}} \frac{D_{\mathrm{G}}}{\tau_{\mathrm{G}}} k_{\mathrm{G}} \\ \theta D_{\mathrm{LG}ij} &= \theta D_{\mathrm{LG}ji} = (\alpha_1 - \alpha_2) \frac{u_i u_j}{|\mathbf{u}|} \end{split}$$

In these equations, D_{LGii} are the principal components of the liquid-gas dispersion tensor; D_{LGij} and D_{LGji} are the cross terms; α is the dispersivity (m) where the subscripts "1" and "2" denote longitudinal and transverse dispersivities, respectively; D_m and D_G (m²/d) are molecular diffusion coefficients; and τ_L and τ_G give the tortuosity factors for liquid (water) and gas (air), respectively.

The three solutes — aldicarb, aldicarb sulfoxide, and aldicarb sulfone — have different decay terms, R_{Li} , partition coefficients, k_{Pi} , and volatilization constants, k_{Gi} . All of the solutes adsorb to soil particles, but only two of the solutes volatilize; sulfoxide does not.

Model Data

The following table provides data for the fluid-flow model:

VARIABLE	UNIT	DESCRIPTION	UPPER LAYER	LOWER LAYER
$K_{ m s}$	m/d	Saturated hydraulic conductivity	0.298	0.454
$\theta_{\mathbf{s}}$		Porosity/void fraction	0.399	0.339
$\theta_{\mathbf{r}}$		Residual saturation	0.001	0.001
α	m ⁻¹	alpha parameter	1.74	1.39
n		n parameter	1.38	1.60
H_{p0}	m	Pressure head in ring	0.01	
$H_{p,\mathrm{init}}$	m	Initial pressure head	-(z+1.2) -0.2(z+0.4)	-(z+1.2)

VARIABLE	UNITS	DESCRIPTION	VALUE
ρ_b	kg/m ³	Bulk density	1400
k_p	m ³ /kg	Partition coefficient	0.0001
D_m	m ² /d	Coefficient of molecular diffusion	0.00374
α_r	m	Longitudinal dispersivity	0.005
α_z	m	Transverse dispersivity	0.001
ϕ_L	d-I	Decay rate in liquid	0.05
ϕ_P	d-I	Decay rate on solid	0.01
c_0	mol/m ³	Solute concentration in ring	1.0

The inputs needed for the solute-transport model are:

Results and Discussion

First, review the results of the perfectly mixed reactor model. Figure 3 shows the concentration profiles of aldicarb and all of its decay products as well as the sum of the three most toxic species — aldicarb, aldicarb sulfoxide, and aldicarb sulfone (see Figure 1 for LD_{50} values). Only small amounts of aldicarb remain after 10 days. Considering the

summed-up contributions, contamination levels clearly remain high even after several months.



Figure 3: Concentration profiles as reactions occur during a 100 day time period.

The following results come from the space- and time-dependent model setup. Figure 4 shows the fluid flow in soil after 0.3 days (left) and 1.0 day (right). The plots illustrate the

wetting of the soil with time. As indicated by the arrows, the fluid velocities are relatively high beneath the ponded water.



Figure 4: The effective saturation (surface plot), pressure head (contours), and flow velocity (arrows) in a variably saturated soil after 0.3 days (left) and 1 day (right).

Figure 5 through Figure 7 show the concentration distributions of aldicarb and the equally toxic aldicarb sulfoxide after 1, 5, and 10 days of infiltration. Consistent with the evolving flow field, the main direction of transport is in the vertical direction.



Figure 5: Concentration of aldicarb (left) and aldicarb sulfoxide (right) after 1 day (note the differing color ranges).



Figure 6: Concentration of aldicarb (left) and aldicarb sulfoxide (right) after 5 days.



Figure 7: Concentration of aldicarb (left) and aldicarb sulfoxide (right) after 10 days.

The distribution of aldicarb has clearly reached steady-state conditions after 10 days, a time frame that was also predicted by the ideal reactor model (see Figure 3). Results also show that the soil contamination is rather local with respect to the aldicarb source. The aldicarb sulfoxide, on the other hand, can be expected to affect a considerably larger soil volume for a significantly longer time.

Notes About the COMSOL Implementation

This model makes use of the **Infinite Element Domain** feature. It performs a coordinate scaling to the selected domain such that boundary conditions on the outside of the infinite element layer are effectively applied at a very large distance. Therefore unwanted effects of artificial boundary conditions on the region of interest are suppressed. This allows to model details in an area which is actually very large or infinite.

References

1. J. Bear, Hydraulics of Groundwater, McGraw Hill, 1978.

2. M.Th. van Genuchten, "A closed-form equation for predicting the hydraulic of conductivity of unsaturated soils," *Soil Sci. Soc. Am. J.*, vol. 44, pp. 892–898, 1980.

Application Library path: Subsurface_Flow_Module/Solute_Transport/ pesticide_transport

Modeling Instructions

From the File menu, choose New.

NEW

In the New window, click 🙆 Model Wizard.

MODEL WIZARD

- I In the Model Wizard window, click 🚈 2D Axisymmetric.
- 2 In the Select Physics tree, select Mathematics>ODE and DAE Interfaces> Global ODEs and DAEs (ge).
- 3 Click Add.
- 4 Click 🔿 Study.
- 5 In the Select Study tree, select General Studies>Time Dependent.
- 6 Click **M** Done.

GLOBAL DEFINITIONS

Load the rate constants from file.

Rate constants

- I In the Model Builder window, under Global Definitions click Parameters I.
- 2 In the Settings window for Parameters, locate the Parameters section.
- 3 Click 📂 Load from File.
- 4 Browse to the model's Application Libraries folder and double-click the file pesticide_transport_parameters_1.txt.

5 In the Label text field, type Rate constants.

First consider the aldicarb decomposition kinetics in the water pond treated as a perfectly mixed system.

GLOBAL ODES AND DAES (GE)

Global Equations 1 (ODE1)

Read in a set of the equations defining the reactions.

- I In the Model Builder window, under Component I (compl)>Global ODEs and DAEs (ge) click Global Equations I (ODEI).
- 2 In the Settings window for Global Equations, locate the Global Equations section.
- 3 Click 📂 Load from File.
- **4** Browse to the model's Application Libraries folder and double-click the file pesticide_transport_reactions.txt.
- 5 Locate the Units section. Click **Select Dependent Variable Quantity**.
- 6 In the Physical Quantity dialog box, type concentration in the text field.
- 7 Click 🔫 Filter.
- 8 In the tree, select General>Concentration (mol/m^3).
- 9 Click OK.
- **IO** In the **Settings** window for **Global Equations**, locate the **Units** section.
- II Click Select Source Term Quantity.
- 12 In the Physical Quantity dialog box, type reactionrate in the text field.

I3 Click **----- Filter**.

- I4 In the tree, select Transport>Reaction rate (mol/(m^3*s)).
- I5 Click OK.

STUDY I

Step 1: Time Dependent

- I In the Model Builder window, under Study I click Step I: Time Dependent.
- 2 In the Settings window for Time Dependent, locate the Study Settings section.
- 3 From the Time unit list, choose d.
- 4 In the **Output times** text field, type range(0,1,100).
- **5** In the **Home** toolbar, click **= Compute**.

RESULTS

Concentration of species (100 days)

A plot showing the time dependent concentration is added automatically. Edit this plot to reproduce Figure 3 with the following steps:

- I In the Settings window for ID Plot Group, type Concentration of species (100 days) in the Label text field.
- 2 Click to expand the Title section. From the Title type list, choose None.
- 3 Locate the Plot Settings section.
- 4 Select the y-axis label check box. In the associated text field, type Concentration (mol/m³).

Global I

- I In the Model Builder window, expand the Concentration of species (100 days) node, then click Global I.
- 2 In the Settings window for Global, locate the x-Axis Data section.
- 3 From the Parameter list, choose Expression.
- 4 In the Expression text field, type t.
- **5** From the **Unit** list, choose **d**.

The concentration for all species is added automatically. Add the expression for the sum of the three most toxic species to the table.

6 Locate the y-Axis Data section. In the table, enter the following settings:

Expression	Unit	Description		
cpm_a+cpm_asn+cpm_asx	mol/m^3	Sum of most toxic species		

7 In the Concentration of species (100 days) toolbar, click 💽 Plot.

Now solve the time- and space-dependent transport and reaction problem in the soil.

ADD PHYSICS

- I In the Home toolbar, click 🙀 Add Physics to open the Add Physics window.
- 2 Go to the Add Physics window.
- 3 In the tree, select Fluid Flow>Porous Media and Subsurface Flow>Richards' Equation (dl).
- 4 Find the Physics interfaces in study subsection. In the table, clear the Solve check box for Study 1.
- 5 Click Add to Component I in the window toolbar.

- 6 In the tree, select Chemical Species Transport> Transport of Diluted Species in Porous Media (tds).
- 7 Click to expand the Dependent Variables section. In the Number of species text field, type3.
- 8 In the Concentrations (mol/m³) table, enter the following settings:

c_a

c_asx

c_asn

9 In the table, clear the Solve check box for Study I.

IO Click **Add to Component I** in the window toolbar.

II In the Home toolbar, click 🙀 Add Physics to close the Add Physics window.

ADD STUDY

- I In the Home toolbar, click \sim Add Study to open the Add Study window.
- 2 Go to the Add Study window.
- **3** Find the **Physics interfaces in study** subsection. In the table, clear the **Solve** check box for **Global ODEs and DAEs (ge)**.
- 4 Find the Studies subsection. In the Select Study tree, select General Studies> Time Dependent.
- 5 Click Add Study in the window toolbar.
- 6 In the Home toolbar, click \sim_{1}° Add Study to close the Add Study window.

STUDY 2

Step 1: Time Dependent Load the parameters defining the material properties and the geometry from file.

GLOBAL DEFINITIONS

Transport parameters

- I In the Home toolbar, click Pi Parameters and choose Add>Parameters.
- 2 In the Settings window for Parameters, locate the Parameters section.
- 3 Click 📂 Load from File.
- **4** Browse to the model's Application Libraries folder and double-click the file pesticide_transport_parameters_2.txt.

5 In the Label text field, type Transport parameters.

GEOMETRY I

The modeling domain is made up of the two permeable soil layers, each of which is represented by a rectangular domain in 2D axisymmetry.

Rectangle 1 (r1)

- I In the **Geometry** toolbar, click Rectangle.
- 2 In the Settings window for Rectangle, locate the Size and Shape section.
- **3** In the **Width** text field, type **1.5**.
- 4 In the **Height** text field, type 0.9.
- 5 Locate the **Position** section. In the z text field, type -1.3.
- 6 Click to expand the Layers section. Select the Layers to the right check box.
- 7 Clear the Layers on bottom check box.
- 8 In the table, enter the following settings:

Layer name	Thickness (m)		
Layer 1	0.25		

This additional layer to the right will later be used to define an **Infinite Element Domain**. Read more about it in the Notes About the COMSOL Implementation section. Proceed with the second soil layer.

9 Right-click Rectangle I (rI) and choose Duplicate.

Rectangle 2 (r2)

- I In the Model Builder window, click Rectangle 2 (r2).
- 2 In the Settings window for Rectangle, locate the Size and Shape section.
- 3 In the **Height** text field, type 0.4.
- 4 Locate the **Position** section. In the **z** text field, type -0.4.

To finish the model geometry, add a point on the top boundary marking the pond's outer rim.

Point I (ptI)

- I In the **Geometry** toolbar, click **Point**.
- 2 In the Settings window for Point, locate the Point section.
- **3** In the **r** text field, type **0.25**.
- 4 Click 🟢 Build All Objects.



Now, define the Infinite Element Domain.

DEFINITIONS

Infinite Element Domain 1 (ie1)

- I In the Definitions toolbar, click 🔍 Infinite Element Domain.
- **2** Select Domains 3 and 4 only.
- 3 In the Settings window for Infinite Element Domain, locate the Geometry section.
- 4 From the Type list, choose Cylindrical.

Variables I

I In the **Definitions** toolbar, click $\partial =$ **Local Variables**.

Load the rate expressions from file.

- 2 In the Settings window for Variables, locate the Variables section.
- **3** Click **b** Load from File.
- 4 Browse to the model's Application Libraries folder and double-click the file pesticide_transport_variables.txt.

RICHARDS' EQUATION (DL)

Begin by specifying the properties for the bottom soil layer in the default **Unsaturated Porous Medium** node, then duplicate this node and modify the domain selection and properties to match the top layer.

Unsaturated Porous Medium I

- I In the Model Builder window, under Component I (compl)>Richards' Equation (dl) click Unsaturated Porous Medium I.
- **2** In the **Settings** window for **Unsaturated Porous Medium**, locate the **Porous Medium** section.
- 3 From the Storage model list, choose User defined. In the S_p text field, type Sp_1.

Porous Matrix I

- I In the Model Builder window, click Porous Matrix I.
- 2 In the Settings window for Porous Matrix, locate the Matrix Properties section.
- 3 From the Permeability model list, choose Hydraulic conductivity.
- **4** In the K_s text field, type Ks_1.
- **5** Locate the **Retention Model** section. In the α text field, type alpha_1.
- **6** In the *n* text field, type n_1.
- 7 In the θ_r text field, type thetar_1.

Unsaturated Porous Medium 1

In the Model Builder window, right-click Unsaturated Porous Medium I and choose Duplicate.

Unsaturated Porous Medium 2

- I In the Model Builder window, click Unsaturated Porous Medium 2.
- **2** Select Domains 2 and 4 only.
- **3** In the **Settings** window for **Unsaturated Porous Medium**, locate the **Porous Medium** section.
- 4 In the S_p text field, type Sp_2.

Porous Matrix I

- I In the Model Builder window, expand the Unsaturated Porous Medium 2 node, then click Porous Matrix I.
- 2 In the Settings window for Porous Matrix, locate the Matrix Properties section.
- **3** In the K_s text field, type Ks_2.

- **4** Locate the **Retention Model** section. In the α text field, type alpha_2.
- **5** In the *n* text field, type n_2.
- **6** In the θ_r text field, type thetar_2.

Gravity I

- I In the Model Builder window, under Component I (compl)>Richards' Equation (dl) click Gravity I.
- 2 In the Settings window for Gravity, locate the Gravity section.
- 3 From the Specify list, choose Elevation.

Initial Values 1

- I In the Model Builder window, click Initial Values I.
- 2 In the Settings window for Initial Values, locate the Initial Values section.
- 3 Click the **Pressure head** button.
- 4 In the H_p text field, type (z+1.2).
- 5 Right-click Initial Values I and choose Duplicate.

Initial Values 2

- I In the Model Builder window, click Initial Values 2.
- **2** Select Domains 2 and 4 only.
- 3 In the Settings window for Initial Values, locate the Initial Values section.
- **4** In the H_p text field, type (z+1.2)-0.2*(z+0.4).

Pressure Head I

- I In the Physics toolbar, click Boundaries and choose Pressure Head.
- **2** Select Boundary 5 only.
- 3 In the Settings window for Pressure Head, locate the Pressure Head section.
- **4** In the H_{p0} text field, type Hp0.

Pervious Layer 1

- I In the Physics toolbar, click Boundaries and choose Pervious Layer.
- **2** Select Boundaries 2 and 8 only.
- 3 In the Settings window for Pervious Layer, locate the Pervious Layer section.
- **4** In the $H_{\rm b}$ text field, type -2.
- **5** In the $R_{\rm b}$ text field, type 1/5[d].

TRANSPORT OF DILUTED SPECIES IN POROUS MEDIA (TDS)

Now, set up the transport equation for an unsaturated porous medium, accounting for adsorption, dispersion and volatilization.

I In the Model Builder window, under Component I (comp1) click Transport of Diluted Species in Porous Media (tds).

Unsaturated Porous Medium I

- I In the Physics toolbar, click 🔵 Domains and choose Unsaturated Porous Medium.
- **2** In the **Settings** window for **Unsaturated Porous Medium**, locate the **Domain Selection** section.
- **3** From the Selection list, choose All domains.

Liquid I

- I In the Model Builder window, click Liquid I.
- 2 In the Settings window for Liquid, locate the Saturation section.
- 3 From the list, choose Liquid volume fraction.
- **4** In the θ_1 text field, type dl.theta_1.

This corresponds to the liquid volume fraction.

- 5 From the Liquid fraction time change list, choose Time change in pressure head.
- **6** From the $dH_{\rm p}/dt$ list, choose Time change in pressure head (dl).
- 7 In the $C_{\rm m}$ text field, type dl.Cm.
- 8 Locate the Convection section. From the u list, choose Total Darcy velocity field (dl).

The diffusion coefficients are the same for all species.

9 Locate the Diffusion section. In the D_{L,ca} text field, type D1. Also type D1 in the text fields for the liquid diffusion coefficients of aldicarb sulfoxide and aldicarb sulfone.

Gas I

Again, the diffusion coefficients are the same for all species.

- I In the Model Builder window, click Gas I.
- 2 In the Settings window for Gas, locate the Diffusion section.
- **3** In the $D_{G,ca}$ text field, type Dg. Also type Dg in the text fields for the gas diffusion coefficients of aldicarb sulfoxide and aldicarb sulfone.
- **4** Locate the **Volatilization** section. In the $k_{G,ca}$ text field, type kg_a.
- **5** In the $k_{G,casn}$ text field, type kg_asn.

Unsaturated Porous Medium I

In the Model Builder window, click Unsaturated Porous Medium I.

Adsorption 1

- I In the Physics toolbar, click Attributes and choose Adsorption.
- 2 In the Settings window for Adsorption, locate the Adsorption section.
- **3** From the Adsorption isotherm list, choose User defined.
- 4 Select the **Species c_a** check box.
- **5** In the $c_{P,ca}$ text field, type kp_a*c_a.
- 6 Select the **Species c_asx** check box.
- 7 In the $c_{P,casx}$ text field, type kp_asx*c_asx.
- 8 Select the **Species c_asn** check box.
- **9** In the $c_{\text{P,casn}}$ text field, type kp_asn*c_asn.

Unsaturated Porous Medium I

In the Model Builder window, click Unsaturated Porous Medium I.

Dispersion I

- I In the Physics toolbar, click Attributes and choose Dispersion.
- 2 In the Settings window for Dispersion, locate the Dispersion section.
- 3 From the Dispersion tensor list, choose Dispersivity.
- **4** From the **Dispersivity model** list, choose **Transverse isotropic**.
- **5** In the α table, enter the following settings:

alphar

alphaz

Reactions I

- I In the **Physics** toolbar, click **Domains** and choose **Reactions**.
- 2 In the Settings window for Reactions, locate the Domain Selection section.
- 3 From the Selection list, choose All domains.
- 4 Locate the **Reaction Rates** section. In the R_{ca} text field, type dl.theta_l*(-r_1-r_3).
- 5 In the R_{casx} text field, type dl.theta_l*(r_1-r_2-r_4).
- 6 In the R_{casn} text field, type dl.theta_l*(r_2-r_5).

Outflow I

I In the Physics toolbar, click — Boundaries and choose Outflow.

2 Select Boundaries 2, 8, 12, and 13 only. This corresponds to the right and bottom boundaries.

Concentration 1

- I In the Physics toolbar, click Boundaries and choose Concentration.
- 2 Select Boundary 5 only.
- 3 In the Settings window for Concentration, locate the Concentration section.
- 4 Select the **Species c_a** check box.
- **5** In the $c_{0,ca}$ text field, type c0.
- 6 Select the **Species c_asx** check box.
- 7 Select the Species c_asn check box.

Volatilization 1

- I In the Physics toolbar, click Boundaries and choose Volatilization.
- **2** Select Boundaries 6 and 11 only.
- 3 In the Settings window for Volatilization, locate the Volatilization section.
- **4** In the h_c text field, type Dg/d_s.
- **5** Select the **Species c_a** check box.
- 6 Select the Species c_asn check box.

MATERIALS

Some required material properties have not yet been defined. This is indicated by a small red cross at the material node. Continue as follows to add the missing properties.

Porous Material: Lower Layer

- I In the Model Builder window, under Component I (comp1) right-click Materials and choose More Materials>Porous Material.
- 2 In the Settings window for Porous Material, type Porous Material: Lower Layer in the Label text field.
- 3 Locate the Phase-Specific Properties section. Click 🙀 Add Required Phase Nodes.

Fluid I (pmat1.fluid1)

- I In the Model Builder window, click Fluid I (pmatl.fluidI).
- 2 In the Settings window for Fluid, locate the Material Contents section.

3 In the table, enter the following settings:

Property	Variable	Value	Unit	Property group
Density	rho	rhof	kg/m³	Basic

Solid | (pmat1.solid1)

- I In the Model Builder window, click Solid I (pmat1.solid1).
- 2 In the Settings window for Solid, locate the Solid Properties section.
- **3** In the θ_s text field, type 1-poro_1.

4 Locate the Material Contents section. In the table, enter the following settings:

Property	Variable	Value	Unit	Property group
Density	rho	rhob	kg/m³	Basic

Porous Material: Lower Layer (pmat1)

In the Model Builder window, right-click Porous Material: Lower Layer (pmatl) and choose Duplicate.

Porous Material: Upper Layer

- In the Model Builder window, under Component I (compl)>Materials click
 Porous Material: Lower Layer I (pmat2).
- **2** In the **Settings** window for **Porous Material**, type **Porous Material**: Upper Layer in the **Label** text field.
- **3** Select Domains 2 and 4 only.
- Solid I (pmat2.solid I)
- I In the Model Builder window, expand the Component I (compl)>Materials> Porous Material: Upper Layer (pmat2) node, then click Solid I (pmat2.solidI).
- 2 In the Settings window for Solid, locate the Solid Properties section.
- **3** In the θ_s text field, type 1-poro_2.

MESH I

Using a mapped mesh is a good idea for this geometry. It uses less mesh elements while keeping the accuracy compared to using a triangular mesh with the same mesh size.

- I In the Model Builder window, under Component I (compl) click Mesh I.
- 2 In the Settings window for Mesh, locate the Physics-Controlled Mesh section.
- **3** From the **Element size** list, choose **Finer**.

Mapped I

In the **Mesh** toolbar, click **Mapped**.

Size 1

- I Right-click Mapped I and choose Size.
- 2 In the Settings window for Size, locate the Geometric Entity Selection section.
- 3 From the Geometric entity level list, choose Domain.
- **4** Select Domain 2 only.
- 5 Locate the **Element Size** section. Click the **Custom** button.
- 6 Locate the Element Size Parameters section.
- 7 Select the Maximum element size check box. In the associated text field, type 0.02.



8 In the Model Builder window, right-click Mesh I and choose Build All.

STUDY 2

Step 1: Time Dependent

- I In the Model Builder window, under Study 2 click Step I: Time Dependent.
- 2 In the Settings window for Time Dependent, locate the Study Settings section.
- **3** From the **Time unit** list, choose **d**.
- 4 In the **Output times** text field, type range(0,0.1,0.9) range(1,1,10).
- **5** In the **Home** toolbar, click **= Compute**.

RESULTS

Study 2/Solution 2 (sol2)

Flownet, pressure, and concentration plots are created per default. Pressure and concentration are also visualized on a revolved 3D geometry. Visualizing the results on the infinite element domains does not add value to the plots. Focus on the region close to the source and therefore hide the infinite element domains from the plots with the following steps.

Selection

- I In the Model Builder window, expand the Results>Datasets node.
- 2 Right-click Study 2/Solution 2 (sol2) and choose Selection.
- 3 In the Settings window for Selection, locate the Geometric Entity Selection section.
- 4 From the Geometric entity level list, choose Domain.
- **5** Select Domains 1 and 2 only.

Proceed as follows to plot the effective saturation, pressure head, and velocity field at different times.

A plot for the effective saturation is available as a predefined plot.

ADD PREDEFINED PLOT

- I In the Home toolbar, click 📑 Windows and choose Add Predefined Plot.
- 2 Go to the Add Predefined Plot window.
- 3 In the tree, select Study 2/Solution 2 (sol2)>Richards' Equation>Effective Saturation (dl).
- 4 Click Add Plot in the window toolbar.
- 5 In the Home toolbar, click Add Predefined Plot to close the Add Predefined Plot window.

RESULTS

Effective Saturation (dl)

- I In the Model Builder window, click Effective Saturation (dl).
- 2 In the Settings window for 2D Plot Group, locate the Data section.
- 3 From the Time (d) list, choose 0.3.

Arrow Surface 1

Right-click Effective Saturation (dl) and choose Arrow Surface.

Contour I

- I In the Settings window for Contour, locate the Levels section.
- 2 In the Total levels text field, type 15.
- **3** In the Effective Saturation (dl) toolbar, click **I** Plot.

Arrow Surface 1

- I In the Model Builder window, click Arrow Surface I.
- 2 In the Settings window for Arrow Surface, locate the Coloring and Style section.
- 3 From the Color list, choose Black.
- **4** In the Effective Saturation (dl) toolbar, click **O** Plot.

Compare the plot in the Graphics window with that in the left panel of Figure 4.

Effective Saturation (dl)

- I In the Model Builder window, click Effective Saturation (dl).
- 2 In the Settings window for 2D Plot Group, locate the Data section.
- 3 From the Time (d) list, choose I.
- **4** In the **Effective Saturation (dl)** toolbar, click **O Plot**.

Compare with the right panel of Figure 4.

The fourth and fifth default plots show the solute concentration after 10 days. Follow the steps below to reproduce the plots shown in Figure 5-Figure 7.

Concentration, aldicarb

- I In the Model Builder window, under Results click Concentration, a (tds).
- 2 In the Settings window for 2D Plot Group, type Concentration, aldicarb in the Label text field.

Streamline 1

- I In the Model Builder window, expand the Concentration, aldicarb node.
- 2 Right-click Streamline I and choose Disable.

Surface 1

- I In the Model Builder window, click Surface I.
- 2 In the Settings window for Surface, click to expand the Range section.
- **3** Select the **Manual color range** check box.
- **4** In the **Minimum** text field, type **0**.

These settings give a single color range for all three aldicarb concentration plots.

5 Click Plot Previous to plot the concentration after 9 days. Repeat to plot the concentration after 5 and 1 day.

Concentration, aldicarb sulfoxide

- I In the Model Builder window, under Results click Concentration, asx (tds).
- 2 In the Settings window for 2D Plot Group, type Concentration, aldicarb sulfoxide in the Label text field.

Streamline 1

- I In the Model Builder window, expand the Concentration, aldicarb sulfoxide node.
- 2 Right-click Streamline I and choose Disable.

Surface 1

- I In the Model Builder window, click Surface I.
- 2 In the Settings window for Surface, locate the Range section.
- **3** Select the **Manual color range** check box.
- **4** In the **Minimum** text field, type **0**.
- 5 In the Maximum text field, type 0.5.

Because the maximum concentration for aldicarb sulfoxide is lower, this setting makes better use of the color range.

Concentration, aldicarb sulfoxide

- I In the Model Builder window, click Concentration, aldicarb sulfoxide.
- 2 In the Settings window for 2D Plot Group, locate the Data section.
- 3 From the Time (d) list, choose 5.
- **4** In the **Concentration**, aldicarb sulfoxide toolbar, click **O** Plot.

Repeat to plot the aldicarb sulfoxide concentration after 10 days.

You can also visualize the evolution of the concentration distributions with time through animations. By adding **Animation** features under the **Export** node, you can generate movies in GIF, WebM, and AVI format. Alternatively, add Player features with animations contained inside the model MPH-file as follows:

Concentration, a, 3D (tds)

In the Model Builder window, click Concentration, a, 3D (tds).

Animation 1

In the Concentration, a, 3D (tds) toolbar, click **....** Animation and choose Player.